

Acylation of 2',3',5'-Tri-*O*-acetylguanosine

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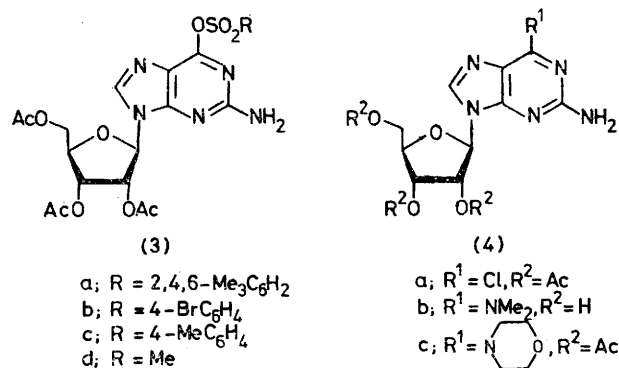
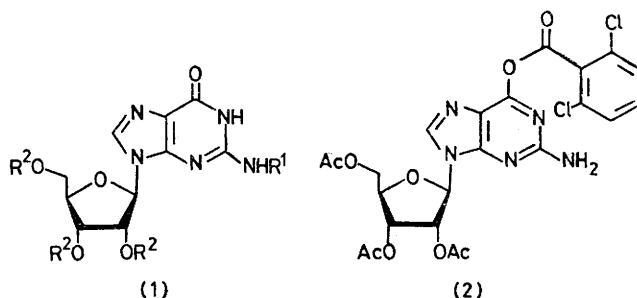
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Summary 2',3',5'-Tri-*O*-acetylguanosine (**1b**) reacts with 2,6-dichlorobenzoyl and mesitylenesulphonyl chlorides to give the corresponding crystalline *O*(6)-acyl derivatives [(**2**) and (**3a**), respectively].

GUANOSINE (**1a**) reacts with acetic anhydride in pyridine solution to give 2',3',5'-tri-*O*-acetylguanosine¹ (**1b**) which then undergoes further acylation on N(2) to give the tetra-acyl derivative² (**1c**). Treatment of guanosine with benzoyl chloride in pyridine solution gives *N*(2),*O*(2'),*O*(3'),*O*(5')-tetrabenzoylguanosine² (**1d**), and (**1b**) similarly undergoes acylation on N(2) when it is treated with *p*-anisoyl chloride under the same conditions. We now report that when (**1b**) is allowed to react with 2,6-dichlorobenzoyl chloride or with arenesulphonyl chlorides in pyridine solution, acylation takes place on *O*(6) of the guanine residue.

Treatment of (**1b**) with an excess of 2,6-dichlorobenzoyl chloride in pyridine solution gives the *O*(6)-aroyl derivative (**2**) as virtually the sole product. The latter compound (**2**) has been isolated pure in 61% yield and obtained as colourless crystals, m.p. 185 °C; its structure has been assigned on the basis of microanalytical and spectroscopic data.† 2',3',5'-Tri-*O*-acetylguanosine¹ (**1b**) reacts similarly with mesitylene-, *p*-bromobenzene-, and toluene-*p*-sulphonyl chlorides to give the corresponding *O*(6)-arenesulphonyl derivatives (**3a**, **3b** and **3c**, respectively). Of these compounds only the first (**3a**) has been obtained crystalline (68% yield, m.p. 141–142 °C). However, the spectroscopic properties of all three compounds closely resemble those of 2-amino-6-chloro-9-β-D-(2',3',5'-tri-*O*-acetylribofuranosyl) purine³ (**4a**) and also those of (**2**). The structure of (**3a**) follows conclusively from its conversion, in nearly quantitative yield, to (**4b**)³ by treatment first with a 3-fold excess of dimethylamine in dioxan-methanol (10:1 v/v) for 1 h at 20 °C followed by deacetylation with methanolic ammonia. Reaction between (**1b**) and methanesulphonyl chloride in

pyridine solution also appears, on the basis of spectroscopic evidence, to give the *O*(6)-mesyl derivative (**3d**) but the latter compound has so far been isolated only in modest yield.‡



† Satisfactory microanalyses have been obtained for all new crystalline compounds described. The most notable feature of the n.m.r. spectrum (CDCl₃) of (**2**) is a broad singlet at δ 5.30 (2H), assignable to the resonance of the NH₂ protons, which disappears on shaking with D₂O. The i.r. spectrum of (**2**) (KBr disc), which has no absorption bands between 1630 and ca. 1740 cm⁻¹, suggests the absence of an *N*-aroyl group; the u.v. spectrum (95% EtOH) of (**2**) exhibits a maximum at 305 nm and is uncharacteristic of an *N*(2)-acyl derivative of guanosine (see ref. 2).

‡ We previously reported (P. K. Bridson, W. Markiewicz, and C. B. Reese, *J.C.S. Chem. Comm.*, 1977, 447) that (**1d**) reacts with methanesulphonyl chloride and triethylamine in dichloromethane solution to give its *O*(6)-mesyl derivative which may be isolated as a pure crystalline solid in 75% yield.

It is as yet unclear why the guanine residue of (**1b**) is attacked by some acylating agents on N(2) and by others on O(6). However, the crystalline O(6)-acyl guanosine derivatives [(**2**) and (**3a**)] are potentially useful synthetic intermediates. The mesitylenesulphonyl derivative (**3a**) is, as indicated above, particularly susceptible to nucleophilic substitution at C(6); indeed, it reacts with morpholine in dioxan solution to give (**4c**) at *ca.* twice the rate of (**4a**). Both derivatives [(**2**) and (**3a**)] are readily unblocked at O(6)

($t_{\frac{1}{2}}$ = 75 and 70 min, respectively) by treatment with 0.5 M potassium carbonate in water-ethanol (1:1 v/v) at 20 °C; it is therefore possible that they may be used as intermediates in the synthesis of otherwise inaccessible guanosine derivatives.

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¹ H. Brederick, *Chem. Ber.*, 1947, **80**, 401.

² C. B. Reese and R. Saffhill, *J.C.S. Perkin I*, 1972, 2937.

³ J. F. Gerster, J. W. Jones, and R. K. Robins, *J. Org. Chem.*, 1963, **28**, 945.